

5.05 2005 Exam 3. April 12, 2005

Instructions: This exam is closed-book. Please write on the exam paper your start and end time and do not exceed 2 h total. Write your name on the exam. Illegible answers will not be credited, so please write clearly.

- Spin-orbit effects in main group chemistry.** Based *in part* on your scrutiny of the Pyykkö paper,¹
 - describe each of the three terms that determine the ¹³C NMR shielding for a compound such as iodomethane, and
 - make a prediction as to what molecule has the largest negative ¹³C NMR shift (δ , ppm) of all, and explain your reasoning.
- The remarkable chemistry of gold.**
 - What is the solvent system used in the synthesis of $[\text{AuXe}_4]^{2+}$ and why is the solvent choice crucial?
 - Write down a balanced chemical equation corresponding to the synthesis of $[\text{AuXe}_4]^{2+}$. Draw structures of all the xenon-containing products with stereochemistry clearly indicated.
 - The synthesis of $[\text{N}(\text{AuPPh}_3)_5]^{2+}$ proceeds in two stages. Write balanced chemical equations corresponding to these stages and discuss briefly the structural chemistry of the key molecules.
 - What interatomic distance is typical of an aurophilic interaction?
 - In no more than one paragraph describe the origin of the aurophilicity phenomenon.
- Noble gas chemistry.** Write down equations showing the product(s) of reaction of 1.0 KrF₂ with each of the following:
 - 2/7 Au.
 - 1/7 I₂.
 - 1/3 Xe.
 - 2/3 B(OTeF₅)₃.
 - 2 SbF₅.
- Noble gas chemistry.** The synthesis of KrF₂ must overcome what two (interrelated) main obstacles? Contrast the situation with the requirements for XeF₂ synthesis.
- Noble gas chemistry.** How does the MO picture for the bonding in KrF₂ contrast with the simple VSEPR approach to its electronic structure?
- Inert pair effect.** Who coined this term and what does it mean?
- Polynitrogen compounds.** In the synthesis of $[\text{N}_5][\text{Sb}_2\text{F}_{11}]$,
 - what material is the reaction vessel made of and,
 - what material is the vacuum line made of?
 - Explain your answers to both (a) and (b).
 - Describe the protocol followed for preparing the reaction vessel prior to carrying out the synthesis.
- Polynitrogen compounds.** Contrast the structures of N_5^+ and N_5^- , and discuss what has been the closest approach to date to synthesizing the latter.

¹ Spend no more than 10-15 minutes on this question. The Pyykkö paper is exempt from closed-book status for the purposes of this exam.

Please refer to the following Pyykkö paper for problem 1:

Kaupp, Martin, Olga L. Malkina, Vladimir G. Malkin, and Pekka Pyykkö. "How Do Spin-Orbit-Induced Heavy-Atom Effects on NMR Chemical Shifts Function? Validation of a Simple Analogy to Spin-Spin Coupling by Density Functional Theory (DFT) Calculations on Some Iodo Compounds." *Chem. Eur. J.* 4, no. 1 (1998): 118-126.