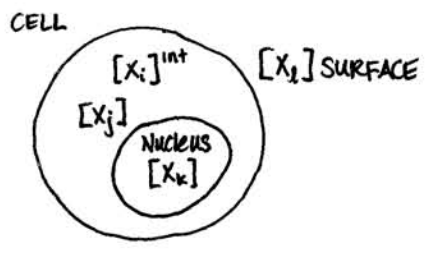


DIFFERENTIAL EQUATION MODEL



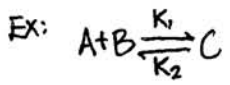
VERY GENERAL SETTING

$$\begin{bmatrix} [x_1] \\ \vdots \\ [x_n] \end{bmatrix} = \vec{x} \quad \vec{u} = \begin{bmatrix} [u_1] \\ \vdots \\ [u_m] \end{bmatrix}$$

time-variation of concentrations

$$\frac{d}{dt}[x_i] = V^+(\vec{x}, \vec{u}) - V^-(\vec{x}, \vec{u})$$

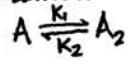
rate of production rate of consumption



$$\begin{aligned} \frac{d}{dt}[C] &= k_1[A][B] - k_2[C] \\ \frac{d}{dt}[A] &= k_2[C] - k_1[A][B] \\ \frac{d}{dt}[B] &= k_2[C] - k_1[A][B] \end{aligned}$$

+ Initial conditions
 $A(0), B(0), C(0)$

Ex: Dimerization



$$\begin{aligned} \frac{d}{dt}[A_2] &= k_1[A][A] - k_2[A_2] \\ \frac{d}{dt}[A] &= 2k_2[A_2] - 2k_1[A][A] \end{aligned}$$

STEADY-STATE (Equilibrium)

Reactions are in balance

$$\frac{d}{dt}[x_i] = 0$$

continuum approximation
 b/c molecules are reacting, but overall the net rate is balanced
 ↓
 Lots of copies of x

Dimerization case:

$$\begin{aligned} k_1[A][A] &= k_2[A_2] \\ [A]^2 &= \frac{k_2}{k_1}[A_2] \\ [A] &= \sqrt{\frac{k_2}{k_1}}[A_2]^{1/2} \end{aligned}$$

Also given:

$$A(0) = A_0 \quad A_2(0) = 0$$

→ initial conditions

Steady-state:

$$[A] + 2[A_2] = A_0$$

CHEMICAL KINETICS FORM

$$\frac{d}{dt}[x_i] = \sum_{j=1}^N \gamma_{ij}^i [x_j] \quad \begin{matrix} \text{destination} \\ \text{source} \end{matrix}$$

(Typical Case $\gamma_i^i < 0$
 degradation process)

$$\frac{d}{dt}[x_i] = \sum_{j=1}^N \gamma_j^i [x_j] + \sum_{j=1}^N \sum_{k=1}^N \gamma_{j,k}^i [x_j][x_k]$$

(Typical Consumption Process $\gamma_{j,k}^i < 0$
 Typical Production Process $\gamma_{j,k}^i > 0$)

$$\begin{aligned} \frac{d}{dt}[x_i] &= \sum_{j=1}^N \gamma_j^i [x_j] + \sum_{j=1}^N \sum_{k=1}^N \gamma_{j,k}^i [x_j][x_k] + \sum_{j=1}^M \alpha_j^i [u_j] + \sum_{j=1}^M \sum_{k=1}^N \alpha_{j,k}^i [u_j][x_k] \\ &\quad + \sum_{j=1}^M \sum_{k=1}^M \beta_{j,k}^i [u_j][u_k] \end{aligned}$$

MULTIPLE INPUTS

$$\begin{aligned} \gamma_i^i < 0 &\Rightarrow \text{models degradation} \\ \gamma_j^i \neq 0 &\Rightarrow \text{models conversion} \quad x_i \xrightleftharpoons[\gamma_j^i]{\gamma_j^i} x_j \\ \gamma_{j,k}^i & \quad x_j + x_k \xrightleftharpoons[\gamma_{j,k}^i]{\gamma_{j,k}^i} x_i \end{aligned}$$

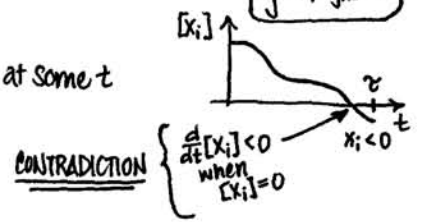
If $[x_i]$ is involved only in reactions that consume $[x_i] \Rightarrow \gamma_i^i \leq 0, \gamma_{i,k}^i \leq 0, \alpha_{j,i}^i \leq 0$
 and reactions that consume $[x_i]$ must include $[x_i] \Rightarrow$

unnecessary for following proof

$$\begin{matrix} j, k \neq i \\ \gamma_j^i \geq 0, \gamma_{j,k}^i \geq 0 \\ \alpha_j^i \geq 0, \alpha_{j,k}^i \geq 0 \end{matrix}$$

Then All $[x_i] \geq 0$ for all t

PROOF: Suppose $[x_i] < 0$ for some i at some t



CONTRADICTION

$$\frac{d}{dt} \vec{X} = A^{(1)} \vec{X} + A^{(2)} \begin{bmatrix} [x_1][x_1] \\ [x_1][x_2] \\ \vdots \\ [x_1][x_n] \\ [x_2][x_1] \\ [x_2][x_2] \\ \vdots \\ [x_2][x_n] \\ [x_3][x_1] \\ \vdots \\ [x_n][x_n] \end{bmatrix}$$

terms $\sum \gamma_j^i [x_j]$
 $\Rightarrow A_{ij}^{(1)} = \gamma_j^i$

terms appear twice ... not a big deal
 N^2 terms

$$A^{(2)} \Rightarrow A_{ij}^{(2)} = \gamma_{j,k}^i$$

$[(j-1)n+k+1]$

KROENEKER PRODUCT:

$$\vec{X} \otimes \vec{X} = \begin{bmatrix} x_1 \vec{X} \\ x_2 \vec{X} \\ x_3 \vec{X} \\ \vdots \\ x_n \vec{X} \end{bmatrix}$$

$$\frac{d}{dt} \vec{X} = A^{(1)} \vec{X} + A^{(2)} \vec{X} \otimes \vec{X} + B \vec{u} + C \vec{u} \otimes \vec{X} + D \vec{u} \otimes \vec{u}$$

$$\begin{bmatrix} u_1 \vec{X} \\ u_2 \vec{X} \\ \vdots \\ u_M \vec{X} \end{bmatrix} = \begin{bmatrix} [u_1][x_1] \\ [u_1][x_2] \\ \vdots \\ [u_1][x_n] \\ [u_2][x_1] \\ \vdots \\ [u_M][x_n] \end{bmatrix}$$